

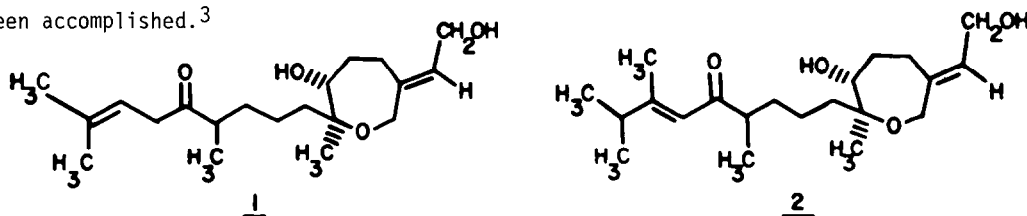
TOTAL SYNTHESIS OF (±) ZOAPATANOL: A STEREOSPECIFIC
SYNTHESIS OF A KEY INTERMEDIATE

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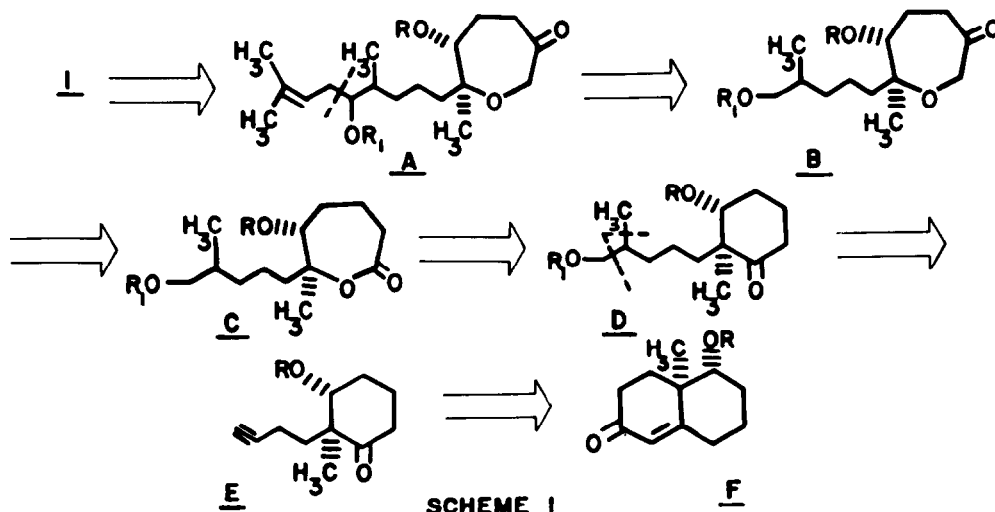
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ABSTRACT: The development of a stereospecific synthesis of a key intermediate in the synthesis of (±)-zoapatanol from the Weiland-Miescher ketone is described.

Zoapatanol (**1**) and montanol (**2**) are two novel biologically active diterpenoids which have been isolated from the zoapatle plant, *Montanoa tomentosa*.² The plant has been used for centuries in the form of a tea to induce menses and labor and terminate early pregnancy. In this and the following communication, we wish to report the total synthesis of (±)-zoapatanol, which also constitutes the synthesis of a diterpenoid containing an oxepane ring. Prior to this work, two other syntheses of (±) zoapatanol detailing totally different approaches had been accomplished.³

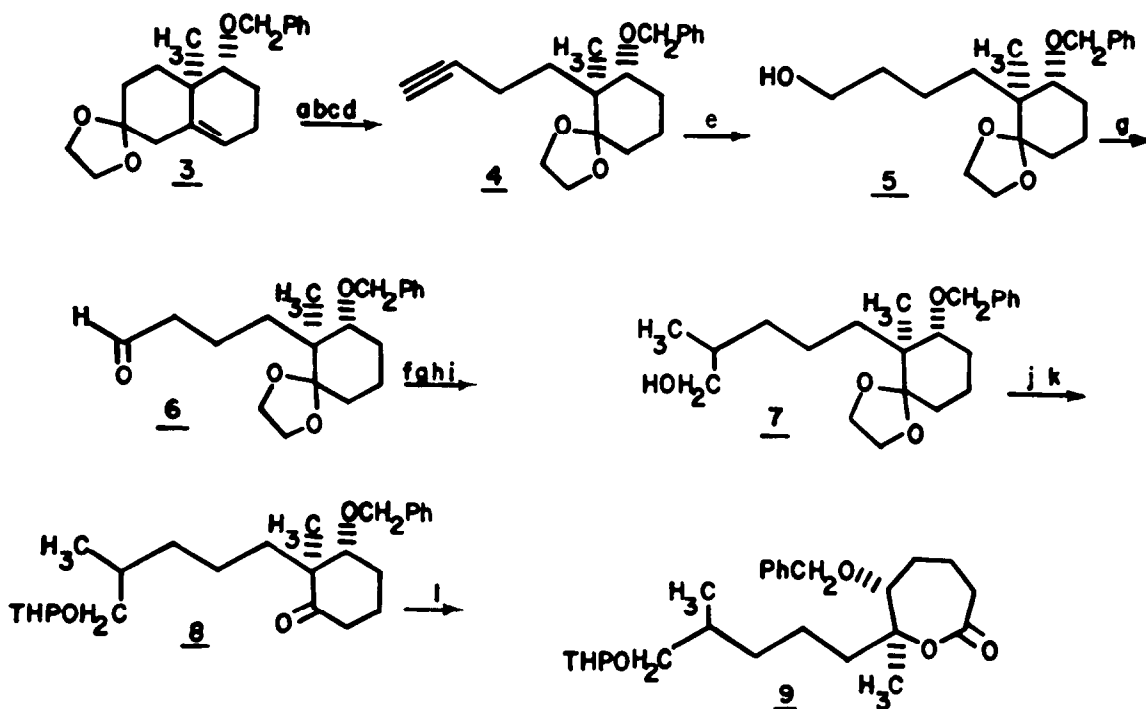


The overall synthetic plan is depicted in the antithetic sequence outlined in Scheme 1. The basic plan entailed the use of bicyclic intermediate **F** to establish and control the stereo-relationship required for the ring geminal alkyl and hydroxyl groups. A key requirement in the synthesis involved conversion of lactone **C** to the β-keto ether **B**. The successful synthesis of a suitable form of **B** is described herein.



SCHEME 1

The requisite starting material, ketal 3⁴ was readily available on a large scale in five steps from the Wieland-Miescher ketone⁵ (Scheme 2). Ketal 3 on acid hydrolysis gave an α,β -unsaturated ketone⁶ (90%),⁷ which upon epoxidation yielded a mixture of keto epoxides (74%). Eschenmoser fragmentation⁸ with *p*-toluenesulfonylhydrazide gave an acetylenic ketone (83%) which was ketalized to give 4 (95%). Hydroboration with 9-BBN^{9,10} (2.1 equivalents) afforded a mixture (85:15 after chromatography) of alcohol 5 and aldehyde 6 (85%). Collins oxidation¹¹ of 5 gave the aldehyde 6 (90%). Treatment of this aldehyde with CH_3Li in Et_2O gave an alcohol (97%), which was oxidized (85%) with Collins reagent.¹¹ The resulting methyl ketone upon treatment with methyltriphenylphosphonium iodide-NaH in DMSO¹² gave the terminal alkene (92%), which was converted to alcohol 7 upon hydroboration^{9,13} with 9-BBN (91%). Hydrolysis of the ketal (99%), followed by protection of the alcohol as its THP derivative (90%) yielded ketone 8 which underwent Baeyer-Villiger oxidation to the desired lactone 9 (70%).

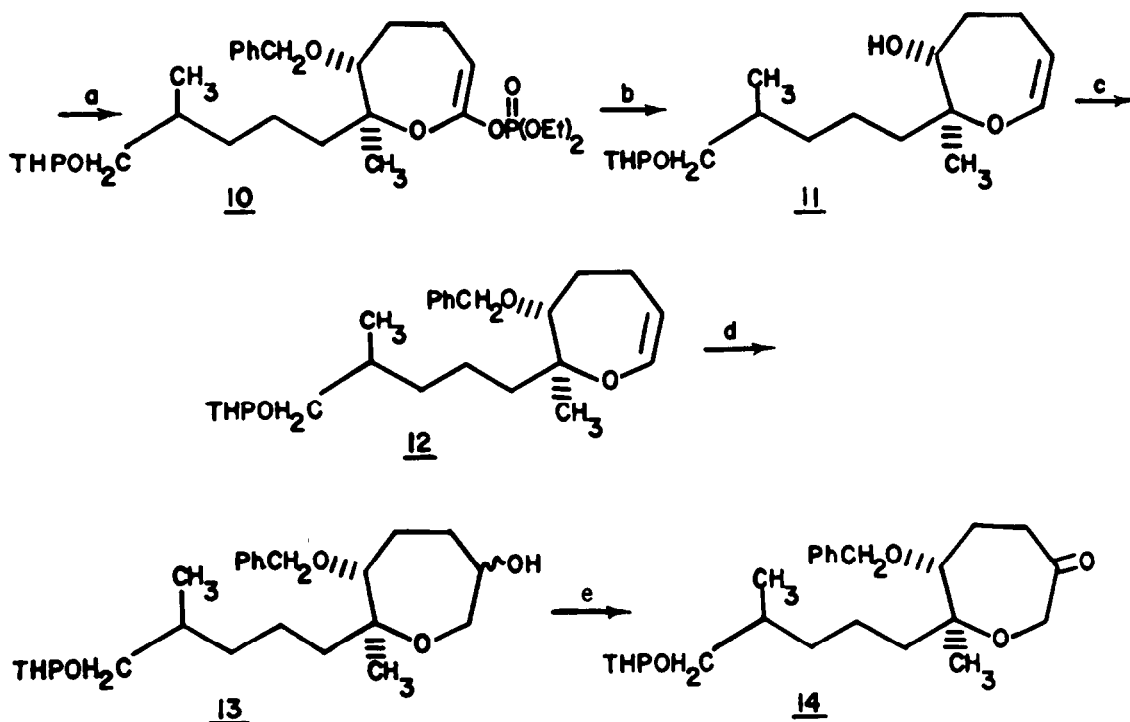


SCHEME 2

- a, $\text{CH}_3\text{COOH}/\text{MeOH}$ (1:1); b, $\text{H}_2\text{O}_2/\text{NaOH}/\text{CH}_3\text{OH}$; c, *p*- $\text{TsNHNH}_2/\text{CH}_3\text{COOH}/\text{CH}_2\text{Cl}_2$;
d, $\text{[OH}^+/\text{TsOH}/\text{C}_6\text{H}_6$; e, 9-BBN/THF; $\text{H}_2\text{O}_2/\text{NaOH}$; f, $\text{CH}_3\text{Li}/\text{Et}_2\text{O}$; g, $\text{CrO}_3/\text{C}_5\text{H}_5\text{N}/\text{CH}_2\text{Cl}_2$;
h, $\text{Ph}_3\text{PCH}_3\text{I}/\text{NaH}/\text{DMSO}$; i, 9-BBN/NaOH/ H_2O_2 ; j, $\text{CH}_3\text{COCH}_3/\text{H}_2\text{O}/0.002\text{N H}_2\text{SO}_4$;
k, DHP/TsOH/ Et_2O ; l, MCPBA/NaOAc/ $\text{CH}_2\text{Cl}_2, \Delta$.

With lactone 9 in hand, we focused our attention on its conversion to the β -keto ether 14 (Scheme 3). After considerable experimentation, an effective, new scheme was developed.¹⁴ Reaction of lactone 9 in THF with LDA at -78°C and treatment of the resulting solution with diethyl chlorophosphate gave the enol phosphate 10 (85%). Reduction of the enol phosphate¹⁵ with Na in a mixture of $\text{NH}_3/\underline{t}\text{-BuOH}/\text{THF}$ led to the desired tetrahydrooxepane 11 (57%), which was benzylated to give the benzyloxy tetrahydrooxepane 12 (98%). Hydroboration of vinyl ether 12 with BH_3 ,^{9,16} afforded a mixture of epimeric alcohols 13 (75%), which upon Collins oxidation¹¹ gave the desired β -keto ether 14 (90%). Elaboration of this β -keto ether 14 to (\pm) zoapatanol is described in the subsequent communication.

SCHEME 3



a, LDA/(EtO) $\overset{\text{O}}{\parallel}$ PCl/THF/TMEDA/HMPA; b, liq $\text{NH}_3/\text{Na}/\underline{t}\text{-BuOH}$; c, $\text{PhCH}_2\text{Br}/\text{NaH}/\text{C}_6\text{H}_6$; d, BH_3/THF ; $\text{H}_2\text{O}_2/\text{NaOH}$; e, $\text{CrO}_3/\text{C}_5\text{H}_5\text{N}/\text{CH}_2\text{Cl}_2$.

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